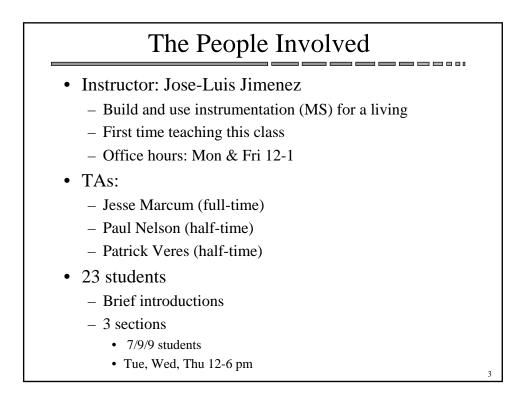
Introduction to CHEM-4181

CU- Boulder CHEM-4181 Instrumental Analysis Laboratory

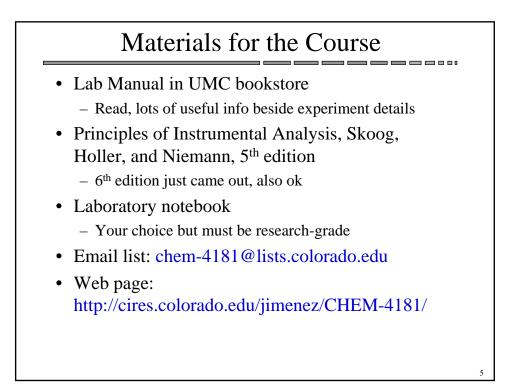
> Prof. Jose-Luis Jimenez Spring 2007

Presentation is posted on course web page – mostly same info as lab manual



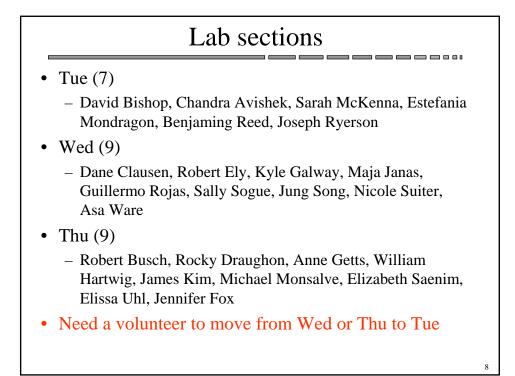
Objectives of the Course

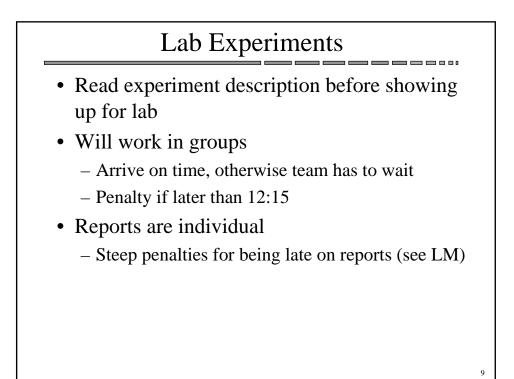
- Review of:
 - Statistical data analysis
 - Scientific writing
 - (These were weak areas based on prior offerings, and you need to do them well for lab reports)
- Fundamental understanding of modern chemical analysis
- Hands-on experience with common instrumental techniques
 - Most basic, a few research-grade instruments. But mostly same procedures as more expensive ones
- Develop critical thinking and evaluation skills within a scientific framework



	Lecture Schedule				
Janua Janua Janua Janua Janua Janua Janua Janua Janua Febri Febri Marc Marc Marc Marc Marc Marc Marc Marc	ure Date ury 17 (W) ary 22 (M) ary 26 F) ary 29 (M) ary 27 (F) aary 5 (M) aary 26 (M) aary 27 (F) aary 26 (M) aary 26 (M) aary 27 (M) by 28-March 2 h 5 (M) th 12 (M) th 14-16 th 16 (F) th 21 (W) h 23 (F) h 26 - 30 2 - April 13 16 (M) 18 & 20 23 (M) 25 (W) 27 (F) 30 (M) 2 (W) 4 (F)	Topic Introduction Data Evaluation and Statistics Scientific Writing Scientific Writing No Class Highlights for Lab Experiments No Class Highlights for Lab Experiments Student Choice Experiment Discussion No Class Student Choice Exp. Approval (in person or e-mail) Student Choice Presentations No Class Spring Break, No Class No Class Crafting a Scientific Presentation No Class Student Choice Presentations St	Ţ		

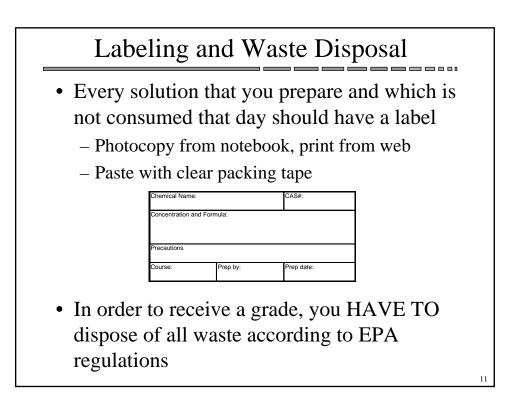
Lab Schedule			
 <u>Week Beginning</u> January 15 January 22 	Experiment/Topic No Lab Introduction, Check In Group Assignments Rotation Sign-Up (groups & individuals)		
 January 29 Feb. 5 – Feb. 26 	Glassware Cleaning Introduction to and Preparation for COD Exp. (E1) Chemical Oxygen Demand (E2) Atomic Absorbance (Pb in soil) (E3) Molecular Absorbance (Phosphorous in water)		
 Mar. 5 – Mar 19 March 26 	(E4) Fluorescence (Oil in water) (E5) FTIR (CO in car exhaust) (E6) GC (PAH's in diesel exhaust) (E7) HPLC (PAH's in cigarette smoke) (E8) Electrochemistry (metal ions in water)		
 March 26 April 2 April 9 April 16 April 23 April 30 	Spring Break – No Lab Student Choice Experiment Student Choice Experiment Finish Student Choice and Check Out No Lab – Student Choice Presentations in Class No Lab – Student Choice Presentations in Class		





A Good Lab Experiment

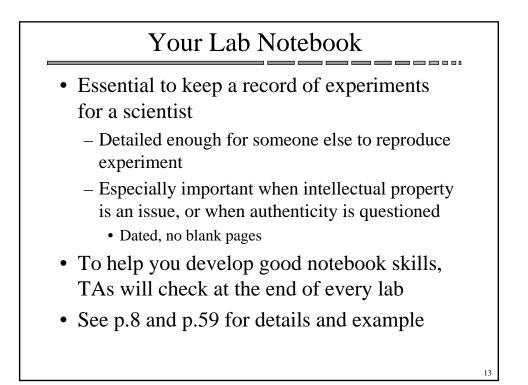
- Clearly Identify the question
- Qualitative vs. quantitative
- Determine most appropriate technique(s)
- Do you need to separate analyte from matrix?
- Acquire a representative sample
- See p. 7-8

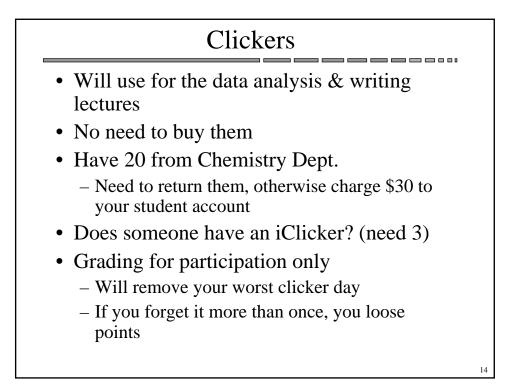


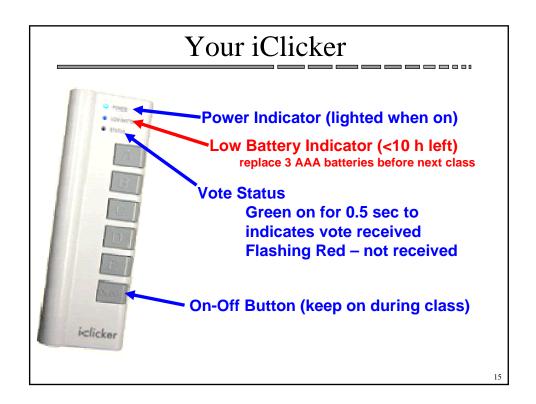
Grading

- Homework: 2 x 30 pts
- Lab reports: 8 x 60 pts
 - Submit to TA within 15 min of starting next lab
 - Steep penalties for being late (see LM)
- Weekly lab evaluations by TA
 - Performance = 8 x 10 pts (0 if late to arrive)
 - Notebook = 8 x 5 pts
- Student choice total = 210!
- Clickers: 30
- Bonus points for helping improve course

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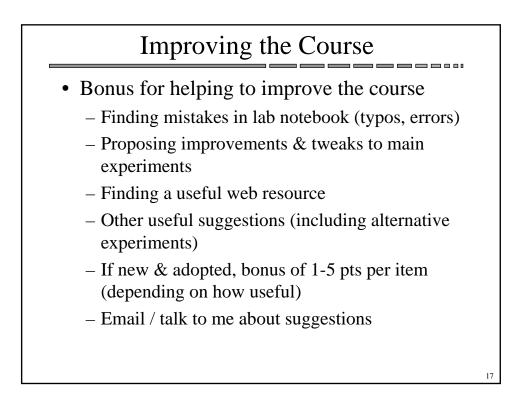




Have you used clickers in a prior class?

- A: Yes, iClickers
- B: Yes, but not iClickers
- C: No
- D: I can't remember

You are NOT locked in to your first choice. Final answer entered will be the one accepted



Student Choice Experiments I

- Normally groups of 2 people (or individual)
- Should be a real problem
 - Environmental, food analysis
 - Has to be done with instruments in class, can't use instruments available to you elsewhere
- Effort in creativity (e.g. caffeine in chocolate done every year)
- Procedures with extensive chemical workup are not appropriate (e.g. 1 page of workup in journal article => too much time)
- Must be planned and approved by March 16
 - Will discuss in class on March 12
 - But you can start thinking now (see example p. 66 of LM)

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