⟨Synapt G2 LC/MS >>

Preparing the samples

1. Prepare the following samples:

Table 3: Sample composition

Sample	Composition
Blank	10:90 acetonitrile:water M.W. acetaminophen (151)
Test Mixture	caffeine (194) sulfadimethoxine (310) reserpine (608) terfenadine (471) (each 2 ng/µL in 10:90 acetonitrile:water)
Calibration	5 mM sodium formate solution (in 30-mL fluidics bottle)
Lock Mass	leucine enkephalin (2 ng/µL in 50:50 acetonitrile:water, in 30-mL fluidics bottle)

NOTE:

Theses samples are the same as those used in the SYNAPT G2 ACQUITY UPLC System performance test during installation. Stock solutions of the test mixture (p/n 700002741) and leucine enkephalin (p/n 700002456) are available from Waters.

Paracetamol

From Wikipedia, the free encyclopedia

Paracetamol INN (/pærəˈsiːtəmol/

or /,pærə'sɛtəmol/), or acetaminophen USAN (♠ i/ə si:tə mɪnəfin/), is a widely used over-thecounter analgesic (pain reliever) and antipyretic (fever reducer). It is commonly used for the relief of headaches, other minor aches and pains, and is a major ingredient in numerous cold and flu remedies. In combination with opioid analgesics, paracetamol can also be used in the management of more severe pain such as post surgical pain and providing palliative care in advanced cancer patients.^[4] The onset of analgesia is approximately 11 minutes after oral administration of paracetamol, [5] and its half-life is 1-4 hours.

While generally safe for use at recommended doses (1,000 mg per single dose and up to 4,000 mg per day for adults, up to 2,000 mg per day if drinking alcohol), [6] acute overdoses of paracetamol can cause potentially fatal liver damage and, in rare individuals, a normal dose can do the same; the risk is heightened by alcohol consumption. Paracetamol toxicity is the foremost cause of acute liver failure in the Western world, and accounts for most drug overdoses in the United States, the United Kingdom, Australia and New Zealand. [7][8][9][10]

It is the active metabolite of phenacetin, once popular as an analgesic and antipyretic in its own right, but unlike phenacetin and its combinations, paracetamol is not considered to be carcinogenic at therapeutic doses. [11] The words acetaminophen (used in the United States, Canada, Japan, South-Korea, Hong Kong, and Iran^[12]) and *paracetamol* (used elsewhere) both come from chemical names for the compound: para-acetylaminophenol and paraacetylaminophenol. In some contexts, it is simply abbreviated as APAP, for acetyl-para-aminophenol.

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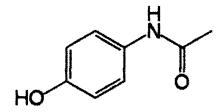
Formula

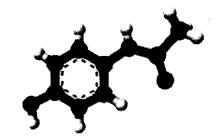
C₈H₉NO₂

Mol. mass

151.17 g/mol

Acetaminophen





Systematic (IUPAC) name

N-(4-hydroxyphenyl)ethanamide N-(4-hydroxyphenyl)acetamide

Clinical data

Trade names Tylenol

AHFS/Drugs.com monograph

MedlinePlus

a681004

Licence data

US FDA:link

Pregnancy cat.

A(AU) B(US) safe

Legal status

Unscheduled (AU) GSL (UK) OTC

(US)

Routes

Oral, rectal, intravenous

Pharmacokinetic data

Bioavailability

~100%

Metabolism

90 to 95% Hepatic

Half-life

1-4 h

Excretion

Renal

CAS number

103-90-2

ATC code

N02BE01

Identifiers

PubChem

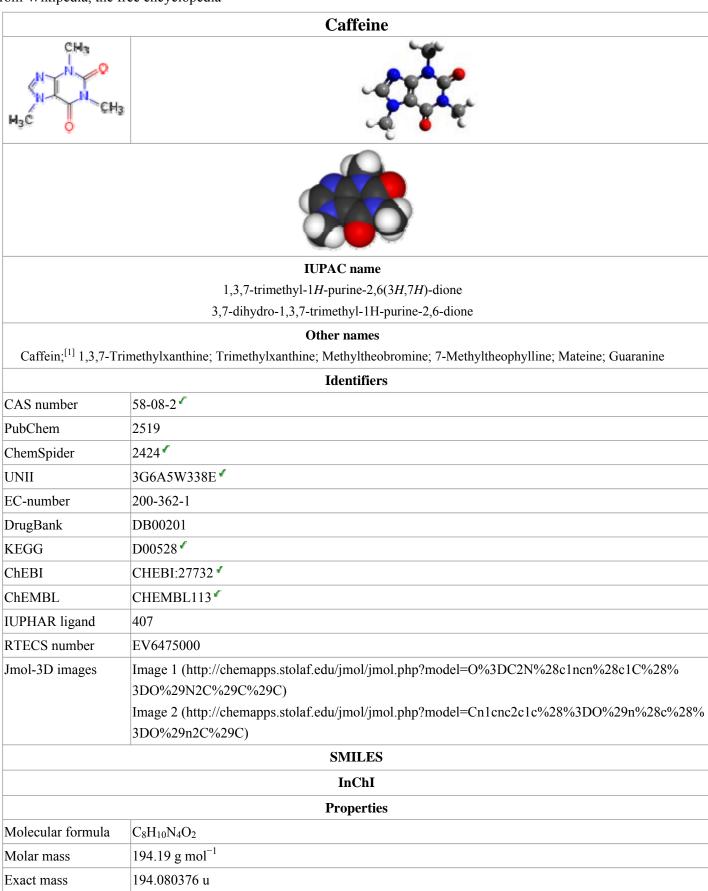
CID 1983

DrugBank

DB00316

Caffeine

From Wikipedia, the free encyclopedia



Sulfadimethoxine

From Wikipedia, the free encyclopedia

Sulfadimethoxine (trade name **Di-Methox**, Albon) is a sulfonamide antibiotic. Albon is produced by Pfizer Animal Health and is available as a oral suspension, tablet or bolus. It is used to treat many infections including treatment of respiratory, urinary tract, enteric, and soft tissue infections. [1]

Sulfadimethoxine Systematic (IUPAC) name 4-amino-*N*-(2,6-dimethoxypyrimidin-4-yl) benzenesulfonamide Clinical data AHFS/Drugs.com FDA Professional Drug Information ? Pregnancy cat. Legal status **Identifiers** 122-11-2 CAS number ATC code J01ED02 QJ01EQ09 QP51AG02 **PubChem** CID 5323 DB06150 **DrugBank** 5132 ChemSpider **UNII** 30CPC5LDEX [◀] D01142 **KEGG** CHEBI:32161 * **ChEBI ChEMBL** CHEMBL62193 * Chemical data **Formula** $C_{12}H_{14}N_4O_4S$ Mol. mass 310.33 g/mol **SMILES** eMolecules & PubChem InChI

(what is this?) (verify)

(http://www.pfizerah.com/PAHimages/compliance_pdfs/US_EN_AO_compliance.pdf) It is most frequently used in veterinary medicine. Sulfadimethoxine inhibits bacterial synthesis of folic acid (pteroylglutamic acid) from para-aminobenzoic acid. It is also commonly used for the treatment of coccidiosis in many species.[2] (http://www.peteducation.com/article.cfm?c=0+1451&aid=1481)

Terfenadine

From Wikipedia, the free encyclopedia

Terfenadine is an antihistamine formerly used for the treatment of allergic conditions. It was brought to market by Hoechst Marion Roussel (now Sanofi-Aventis) and marketed under various brand names including **Seldane** in the United States, **Triludan** in the United Kingdom, and **Teldane** in Australia. According to its manufacturer, terfenadine had been used by over 100 million patients worldwide as of 1990. ^[1] It was superseded by fexofenadine in the 1990s due to the risk of cardiac arrhythmia caused by QT interval prolongation.

Terfenadine is a prodrug, generally completely metabolised to the active form fexofenadine in the liver by the enzyme cytochrome P450 CYP3A4 isoform. Due to its near complete metabolism by the liver immediately after leaving the gut, terfenadine normally is not measurable in the plasma. Terfenadine itself, however, is cardiotoxic at higher doses while its major, active metabolite is not. Toxicity is possible after years of continued use with no previous problems as a result of an interaction with other medications such as erythromycin, or foods like grapefruit. The addition of, or dosage change in, these CYP3A4 inhibitors makes it harder for the body to metabolize and remove terfenadine. In larger plasma concentrations, terfenadine may lead to toxic effects on the heart's rhythm (e.g. ventricular tachycardia and *torsades de pointes*).

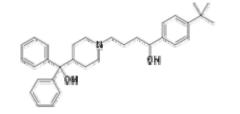
History

In the United States, Seldane was brought to market in 1985 as the first non-sedating antihistamine for the treatment of allergic rhinitis. [1][2] In June 1990, evidence of serious ventricular arrhythmias among those taking Seldane prompted the FDA to issue a report on the risk factors associated with concomitant use of the drug with macrolide antibiotics and ketoconazole. [1] Two months later, the FDA required the manufacturer to send a letter to all physicians, alerting them to the problem; in July 1992 the existing precautions were elevated to a black box warning [1] and the issue attracted mass media attention in reports that people with liver disease or who took ketoconazole, an antifungal agent, or the antibiotic erythromycin, could suffer cardiac arrhythmia if they also took Seldane. [2]

In January 1997, the same month when the U.S. Food and Drug Administration (FDA) had earlier approved a generic version of Seldane made by IVAX Corporation of

Miami, the FDA recommended that terfenadine-containing drugs be removed from the market and that physicians consider alternative medications for their patients. [2] Seldane (and Seldane-D, terfenadine combined

Terfenadine



Systematic (IUPAC) name

 $\label{eq:continuous} (RS)\mbox{-}1\mbox{-}(4\mbox{-}tert\mbox{-}butylphenyl)\mbox{-}4\mbox{-}\{4\mbox{-}[hydroxy(diphenyl)methyl]$} \\ piperidin\mbox{-}1\mbox{-}yl\}\mbox{-}butan\mbox{-}1\mbox{-}ol$

Clinical data

AHFS/Drugs.com Multum Consumer

Information

MedlinePlus a600034

Pregnancy cat. ?

Legal status Withdrawn

Pharmacokinetic data

Protein binding 70%

Half-life 3.5 hours

Identifiers

CAS number 50679-08-8

ATC code R06AX12

PubChem CID 5405

DrugBank APRD00606

ChemSpider 5212

UNII 7BA5G9Y06Q ⁴

KEGG D00521

ChEMBL CHEMBL17157 [₹]

Chemical data

Formula $C_{32}H_{41}NO_2$

Mol. mass 471.673 g/mol

SMILES eMolecules & PubChem

InChI

✗(what is this?) (verify)

Reserpine

From Wikipedia, the free encyclopedia

Reserpine is an indole alkaloid^[2] antipsychotic and antihypertensive drug that has been used for the control of high blood pressure and for the relief of psychotic symptoms, although because of the development of better drugs for these purposes and because of its numerous side-effects, it is rarely used today.^[1] The antihypertensive actions of reserpine are a result of its ability to deplete catecholamines (among other monoamine neurotransmitters) from peripheral sympathetic nerve endings. These substances are normally involved in controlling heart rate, force of cardiac contraction and peripheral resistance.^[3]

Reserpine mediated depletion of monoamine neurotransmitters in the synapses is often cited as evidence to the theory that depletion of the neurotransmitters causes subsequent depression in humans (c.f. monoamine hypothesis). However, this claim is not without controversy; Some have called reserpine-induced depression a myth, while others claim that teas made out of the plant roots containing reserpine have a calming, sedative action that can actually be considered **anti**depressant.^[2] This remains to be demonstrated in the clinic.

Moreover, reserpine has a peripheral action in many parts of the body, resulting in a preponderance of the cholinergic part of the nervous system (GI tract, smooth muscles vessels).

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Mechanism of Action

Reserpine is an irreversible antagonist of VMAT. It acts by blocking the vesicular monoamine transporter VMAT.^[3] This normally transports free norepinephrine, serotonin, and dopamine from the cytoplasm of the presynaptic nerve

Reserpine

Systematic (IUPAC) name

 $methyl-11,17\alpha$ - $dimethoxy-18\beta$ -[(3,4,5-trimethoxybenzoyl)]

Clinical data

AHFS/Drugs.com Consumer Drug Information

MedlinePlus a601107

Licence data US FDA:link

Pregnancy cat. D (fetotoxic)

Legal status Rx-only (some countries

banned/discontinued)

Routes oral

Pharmacokinetic data

Bioavailability 50%

Metabolism gut/liver

Half-life phase 1 = 4.5h,

phase 2 = 271h,

average = 33h

Excretion 62% feces / 8% urine

Identifiers

CAS number 50-55-5

ATC code C02AA02

PubChem CID 5770

DrugBank APRD00472

ChemSpider 5566

UNII 8B1QWR724A

ChEBI CHEBI:28487 [™]

ChEMBL CHEMBL772 [✓]

Chemical data

Formula $C_{33}H_{40}N_2O_9$

Mol. mass 608.68 g/mol

SMILES eMolecules & PubChem